

Single-Atom Skeletal Editing

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Why in News?

Chemists at ETH Zürich developed a milder, direct method for single-atom skeletal editing, allowing the direct swapping of a carbon atom for a nitrogen atom in core organic scaffolds.

Single-Atom "Skeletal Editing"

- **Traditional method** - In drug discovery, modifying the underlying ring framework of a complex molecule traditionally requires destroying the molecule and synthesizing it again from scratch over multiple step-by-step chemical reactions.
- **Skeletal Editing Concept** - Functions like a digital "find-and-replace" tool for molecules, enabling synthetic chemists to delete, insert, or swap single atoms inside a pre-existing, fully assembled molecular skeleton without disturbing surrounding functional groups.
- **Replacing Carbon with Nitrogen** - Swapping a carbon atom (C) for a nitrogen atom (N) alters a drug molecule's polarity, metabolic stability, solubility, and hydrogen-bonding capabilities.
- It often makes the drug more bioavailable or capable of crossing the blood-brain barrier.

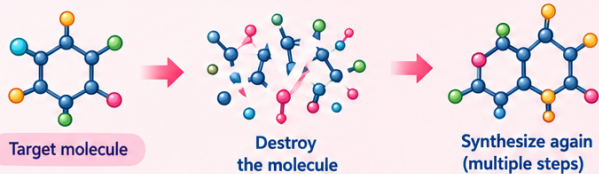


Skeletal Editing in Drug Discovery

Edit molecules. Unlock better medicines.



1 Traditional Method



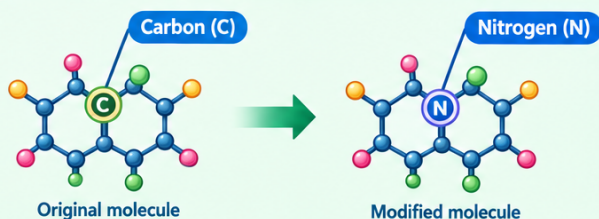
Modifying the ring framework means breaking down the molecule and rebuilding it from scratch through multiple reactions.

2 Skeletal Editing Concept



Like a digital "find-and-replace" tool — delete, insert or swap single atoms without disturbing surrounding functional groups.

3 Replacing Carbon with Nitrogen



What Changes?



Polarity ↑



Metabolic stability ↑



Solubility ↑



Hydrogen-bonding capabilities ↑



The Bigger Benefit

Often makes the drug more bioavailable or capable of crossing the blood-brain barrier.



Smarter chemistry. Better drugs.



Reaction Mechanism

- **Reagents Used** - Operates under mild conditions using a simple, commercially available combination of
 - **Hypervalent Iodine Reagent** - Phenyliodine(III) diacetate (PIDA) as the oxidant.
 - **Nitrogen Source** - Ammonium carbamate ($\text{H}_4\text{N} + \text{H}_2\text{NCOO}$) as a mild, safe nitrogen donor.
- **Cascade Pathway**
 - **Oxidative Cleavage** - The reagent selectively opens the double bond in the core indole ring.
 - **Oxidative Amidation & Hofmann-Type Rearrangement** - Inserts the nitrogen atom while removing a single carbon atom.
 - **Recyclization** - Spontaneously re-closes the ring structure into a benzimidazole scaffold.
- **High Functional Group Tolerance** - Unlike earlier atomic-swap methods requiring harsh reagents (such as ozone or hazardous azides), this mild protocol tolerates sensitive functional groups across highly complex, late-stage drug molecules.

Applications in Pharmaceutical Chemistry

- **Streamlining Drug Discovery** - Enables chemists to perform rapid testing by how adding nitrogen atoms into drug candidates impacts biological efficacy without building new molecules from scratch.
- **Accessing Bio-isosteres** - Facilitates direct interconversion between indoles and benzimidazoles.
- **Late-Stage Diversification** - Successfully demonstrated on complex, drug-like molecules, cutting down research timelines and hazardous chemical waste in pharmaceutical development.

Reference

[The Hindu | 'Milder' way to swap carbon for nitrogen atoms](#)

