

C. V. Raman

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Why in News?

Recently, National Science Day (February 28) is being observed to honour C. V. Raman's discovery of the Raman Effect and his Nobel Prize-winning contribution to physics.

- **Born in** - 7th November 1888, Tiruchirapalli, Tamil Nadu.
- **Parents** - R. Chandrasekhar Iyer (Lecturer in Physics and Mathematics).
- **Early Brilliance** - Won gold medals in Bachelor's and Master's degrees.
- Published first paper at age 18 in *Philosophical Magazine*.
- **Early Career** - Assistant Accountant General, Indian Finance Service (1907), Calcutta.
- **Research Work - Early Phase** - Conducted research at the Indian Association for the Cultivation of Science during spare time.
- **Later Years** - *Optics, crystal dynamics*, diamond structure, colours of flowers, human vision.
- **Academic Career** - Professor of Physics at Calcutta University (Palit Chair) for 15 years.
- Director, Indian Institute of Science (1933-1937).
- Founded Raman Research Institute in 1948.
- **Major Discovery** - Raman Effect (1928).
- **Nobel Prize** - Nobel Prize in Physics (1930), ***first Indian and person of colour to receive it in Physics.***
- **Mentorship** - *Mentored Vikram Sarabhai* in cosmic ray research.
- **Legacy** - Raman Effect used in cancer detection, drug development, and space exploration; inspiration for National Science Day.
- **Death** - 1970.

Raman Effect (Raman Scattering)

- Raman scattering is a spectroscopic technique based on the inelastic scattering of monochromatic light (usually from a laser).
- **Aim** - To study ***vibrational, rotational and other low-frequency*** modes in a system.
- **Principle** - When a photon interacts with a molecule, it transfers energy, causing the molecule to vibrate or rotate.
- The scattered light shows a change in frequency (Raman shift), which is characteristic of the molecule.
- **Types of Raman Scattering - Stokes Raman Scattering** - If the scattered photon has a lower frequency; molecule gains energy.
- **Anti-Stokes Raman Scattering** - If a scattered photon has a higher frequency, the molecule loses energy. (Stokes scattering is more common.)

Reference

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